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AND pOH How to find pH, pOH,  
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*pH (Worksheet) - Chemistry*

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## *pH and pOH Practice Worksheet - ThoughtCo*

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## *pH and Titrations (Worksheet) - Chemistry LibreTexts*

Acid and Base Worksheet - Answers

1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a)  $\text{HNO}_3 +$

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OH-1 (H<sub>2</sub>O + NO<sub>3</sub>-1

## *Acid and Base Worksheet - Answers - Chemistry Made Easy*

Salts such as sodium chloride are made from ions bonded together. 4.(c) pH is the measure of hydrogen ions in acids. A pH equal to 7 is neutral (e.g. of water). A pH less than 7 is acidic. A pH more than 7 is basic. 4.(d) Acids have a pH less than 7, have a sour taste, are corrosive, and are neutralised by bases.

## *Cell Chemistry Answers Worksheet - xcelerate-science.com*

pH is a logarithmic unit of concentration. Specifically,  $\text{pH} = -\log [\text{H}^+]$ ;  $\text{pH} = -\log [\text{H}_3\text{O}^+]$ . It is a scale from 0 to 14. Values  $<7$  are acidic;  $>7$  are basic, and a value  $=7$  is neutral. The pH scale was developed by a



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Belgian brewmaster to control the quality of his product.

*Introduction to Acids and Bases  
(Worksheet) - Chemistry ...*

$[\text{OH}^-] [\text{OH}^-] = 1.00 \times 10^{-7}\text{M}$  The concentrations of  $\text{H}^+$  and  $\text{OH}^-$  are equal, as are the pH and pOH, so the solution must be neutral.

*Calculating pH and pOH worksheet*

This is a collection of chemistry worksheets in pdf format. The answers to the questions are available on separate worksheets so you can fill them out and then check your work. Please feel free to download these to your computer, print them, and use them as hand-outs. Metals, Nonmetals, and Metalloids Worksheet

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Name: Date: pH and pOH About  
Chemistry <http://chemistry.about.com>

Fill in the missing sections: [H+] pH  
[OH-] pOH Acid or Base 1.  $1 \times 10^{-6}$  6  
 $1 \times 10^{-8}$  8 Acid 2.  $1 \times 10^{-}$  ...

*Name: Date: pH and pOH*

Questions are a mixture of short and long answers, similar to an exam, the worksheet requires pupils to explain the process of identifying acids and alkalis, neutralisation and measuring pH. Answers are included so students can peer or self assess their work.

*KS3 Acids and Alkalis Worksheet  
(teacher made)*

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h2opalermo.it*

8. A 1.0 L solution of HCl has a pH =  
1. How many liters of distilled water  
must be added to change the pH to 2?  
9 Liters . 9. 6 g of LiOH is added to  
water to make 500 ml of solution.  
What is the pH? pH = 13.7 . 10. What  
volume of 0.05 M HI is required to  
neutralize 50 ml of 0.01 M Ca(OH)

*Name: Date: Strong Acids and Bases*  
pH practice - Answers 1) What is the  
pH and pOH of a  $1.2 \times 10^{-3}$  HBr  
solution? pH: 2.9 pOH: 11.1 2) What is  
the pH and pOH of a  $2.34 \times 10^{-5}$

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NaOH solution? pOH: 4.6 pH: 9.4 3)  
What is the pH and pOH of a solution  
made by adding water to 15 grams of  
hydroiodic acid until the volume of the  
solution is 2500 mL? pH: 1.6 pOH:  
12.4

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